

# Tosyl ester(s) of $\alpha$ -hydroxy acid(s) – Synthesis and antimicrobial studies

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**Abstract** –Tosyl ester(s) of  $\alpha$ -hydroxy acid(s) were prepared by treating the acid with *p*-toluenesulfonyl chloride (TsCl) in the presence of pyridine. The products were isolated, identified by TLC and characterized by IR, <sup>1</sup>H-NMR and <sup>13</sup>C-NMR spectral techniques. Anti microbial activity of the synthesized compounds were studied by Agar well diffusion method against two bacterial (*Escherichia Coli* and *Staphylococcus Aureus*) and two fungal (*Candida albicans* and *Aspergillus Niger*) species using Ciprofloxacin against the bacterial strains, Nystatin against *Candida albicans* and Amphotericin against *Aspergillus niger* as standard antibiotics. The results infer that the tosyl ester(s) of  $\alpha$ - hydroxy acid(s) are potentially active against bacterial and fungal strains.

**Keywords** – *Tosyl esters,  $\alpha$ -hydroxy acids, Sulfonyl sulfur, TLC technique, Antimicrobial activity, Agar well diffusion method*

## I. INTRODUCTION

Tosyl esters are sulfonyl esters synthesized by treating *p*-toluenesulfonyl chloride (known as tosyl chloride and abbreviated as TsCl) with phenols / aliphatic alcohols or carboxylic acids in the presence of base. Formation of tosyl ester is a nucleophilic substitution reaction in which the oxygen of the alcohol/acid displaces the chloride ion from the tosyl chloride. Base catalyzes the reaction and neutralizes the HCl that formed in the reaction [1, 2].

## II. RELATED WORK

Organo Sulfur compounds find a wide range of chemical, medicinal, biomedical and industrial applications due to their low cost and useful properties such as solubility in aqueous and organic solvents, metal complexing ability, biological compatibility and ease of chemical modification [3-5]. The various methods of preparation of tosyl esters from aliphatic alcohols, aromatic acids, sulfonic acids and phenols have been reported [6-11].

Sulfur containing drugs such as sulfonamides, sulfonyl ureas, 2-thio uracil, 6-mercapto purines etc., have extensive therapeutic usage. Some sulfur containing agrochemicals have useful biological activity. They used as insecticides, herbicides, fungicides and agaricides [12]. Sulfonyl medicines and their antibiotics are commonly used for the treatment of infections occurred by microbes [13]. Various metal complexes of indole acetic acid [14], tosyl esters of

indole acetic acid and naphthalene acetic acid [15] were studied for their antimicrobial activity.

With the development of new applications in medical and other applied areas, there is a growing demand for improved and versatile methods for the synthesis and characterization of sulfonyl derivatives. Hence, we synthesized few tosyl ester(s) of  $\alpha$ - hydroxy acid(s), characterized by spectral techniques and studied their microbiological activities against pathogenic microbes.

## III. METHODOLOGY

### Materials

All the chemicals such as *p*- toluenesulfonyl chloride, glycolic acid, lactic acid, mandelic acid, pyridine and acetonitrile were purified before use by recrystallization or distillation until their physical constants (melting/boiling) agreed well with the literature values [16,17]. FT-IR (KBr) spectra were recorded on Perkin Elmer RXI spectrophotometer and <sup>1</sup>H & <sup>13</sup>C –NMR spectra were recorded on a Bruker AMX 400 MHz & 200 MHz NMR spectrometer respectively using TMS as an internal reference.

### Synthesis of tosyl esters

#### *p*-toluenesulfonyl glycolate (PTSG)

Equal volumes of equimolar solutions of *p*-toluenesulfonyl chloride (25ml, 0.05 mol dm<sup>-3</sup>) and mixture of Glycolic acid – Pyridine (25ml, 0.05 mol dm<sup>-3</sup>) in acetonitrile were mixed

with constant stirring and kept at about 30 °C for overnight. The solid product obtained was treated with ether. The ether layer was separated and kept under vacuum for evaporation to get residue and dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. The solid product obtained was recrystallized from toluene [Yield 0.387g, melting point 106 °C]. The product was identified as *p*-toluenesulfonyl glycolate from IR (KBr), <sup>1</sup>H NMR (200MHz, CDCl<sub>3</sub>) and <sup>13</sup>C- NMR (DMSO-d<sub>6</sub>) spectral data [18].

The same procedure was adopted for the synthesis of *p*-toluenesulfonyl lactate (Yield 0.294g) and *p*-toluenesulfonyl mandalate (Yield 0.302g).

### Characterization of the synthesized compounds

#### Compound I: *p*-toluenesulfonyl glycolate (PTSG)

The spectral data are,

**IR:** 3417 ν<sub>-OH</sub> (H-bonded), 3063 ν<sub>C-H</sub> (aromatic), 2924 ν<sub>C-H</sub> (aliphatic), 1730 ν<sub>C=O</sub>, 1375 ν<sub>S-O</sub> (asy.), 1182 ν<sub>S-O</sub> (sym.), 693 ν<sub>S-O-C</sub> cm<sup>-1</sup>

**<sup>1</sup>H NMR:** 2.31(s, 3H, CH<sub>3</sub>-C<sub>6</sub>H<sub>5</sub>), 3.18 (s, 2H, CH<sub>2</sub>), 7.22-7.20 (d, 2H, C<sub>3</sub>- and C<sub>5</sub>-H, -SO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>CH<sub>3</sub>), 7.68-7.66 (d, 2H, C<sub>2</sub>- and C<sub>6</sub>-H, -SO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>CH<sub>3</sub>), 3.69 (s, 1H, -OH) ppm

**<sup>13</sup>C NMR:** 175, 140,138,129,126, 77 and 22ppm.

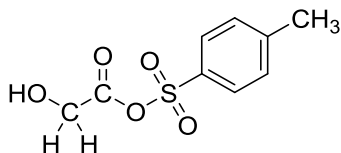


Figure 1: *p*-toluenesulfonyl glycolate

#### Compound II: *p*-toluenesulfonyl lactate (PTSL)

**IR:** 3411 ν<sub>-OH</sub> (H-bonded), 3097 ν<sub>C-H</sub> (aromatic), 2925 ν<sub>C-H</sub> (aliphatic), 1733 ν<sub>C=O</sub>, 1339 ν<sub>S-O</sub>(asy.), 1166 ν<sub>S-O</sub> (sym.), 682 ν<sub>S-O-C</sub> cm<sup>-1</sup>

**<sup>1</sup>H NMR:** 1.20 (d, 3H, CH<sub>3</sub>-CH), 2.30 (s, 3H, CH<sub>3</sub>-C<sub>6</sub>H<sub>5</sub>), 3.11-3.19 (q, 1H, CH<sub>3</sub>-CH), 7.22-7.19 (d, 2H, C<sub>3</sub>- and C<sub>5</sub>-H, -SO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>CH<sub>3</sub>), 7.69-7.66 (d, 2H, C<sub>2</sub>- and C<sub>6</sub>-H, -SO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>CH<sub>3</sub>), 4.05(s, 1H, -OH) ppm

**<sup>13</sup>C NMR:** 174, 140,128,126,125, 76 and 20ppm.

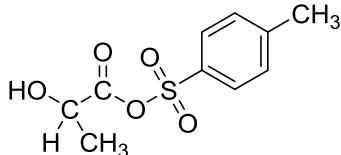


Figure 2: *p*-toluenesulfonyl lactate

#### Compound III: *p*-toluenesulfonyl mandalate (PTSM)

**IR:** 3402 ν<sub>-OH</sub> (H-bonded), 3040 ν<sub>C-H</sub> (aromatic), 2918 ν<sub>C-H</sub> (aliphatic), 1737 ν<sub>C=O</sub>, 1314 ν<sub>S-O</sub> (asy.), 1158 ν<sub>S-O</sub> (sym.), 728 ν<sub>S-O-C</sub> cm<sup>-1</sup>

**<sup>1</sup>H NMR:** 2.28 (s, 3H, CH<sub>3</sub>-C<sub>6</sub>H<sub>5</sub>), 2.48 (s, 1H, HOC<sub>6</sub>H<sub>5</sub>-CH), 7.49 (d, 2H, C<sub>3</sub>- and C<sub>5</sub>-H, -SO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>CH<sub>3</sub>), 7.11 (d,

2H, C<sub>2</sub>- and C<sub>6</sub>-H, -SO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>CH<sub>3</sub>), 7.44-7.25 (m, 5H, C<sub>6</sub>H<sub>5</sub>-CH-) 5.03 (s, 1H, -OH) ppm

**<sup>13</sup>C NMR:** 169, 141,136, 135, 133, 124,123, 122, 121,120, 72 and 16 ppm.

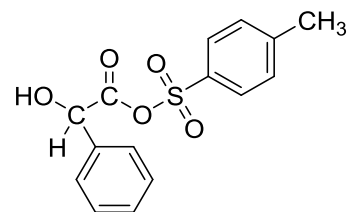


Figure 3: *p*-toluenesulfonyl mandalate

### Antimicrobial Screening

Antimicrobial agent either kills microorganisms or stops their growth. Antimicrobial agents may be antibacterial, antifungal or antiviral. They all have different modes of action by which they act to suppress the infection [19]. Various groups of microorganisms have different extent of sensitivity towards antimicrobials. Microbes may develop drug resistance on continuous use of particular antimicrobial agent. So, antimicrobial studies have significant role to find the activity of particular chemical compound against microorganisms[20- 22].

In the present study Agar well disc diffusion technique has been used for determining the susceptibility of the bacterial (*Escherichia Coli* and *Staphylococcus Aureus*) and fungal (*Candida albicans* and *Aspergillus Niger*) species on the test compounds. Nutrient Agar medium and Rose Bengal Chloramphenicol agar medium were used for culturing bacterial and fungal species respectively. The Agar media were inoculated with test organism and various concentrations (25, 50, 75 and 100 µg/ml in sterile DMSO) of the test solution of the compounds synthesized.

Then a standard antibiotic Ciprofloxacin against the bacterial strains *S. Aureus* and *E. Coli*, Nystatin against *Candida albicans* and Amphotericin against *Aspergillus niger* was placed at the center of the agar plate. The plate with bacterial organism was incubated at 35-37 °C for about 24 hours and the plate with fungal organism was incubated at the same temperature for 48 hours. After incubation the zone of inhibition was measured in mm scale. The results of antibacterial and antifungal screening on the test compounds are furnished in Table 1 and Figure 4 -7.

## IV. RESULTS AND DISCUSSION

Tosyl esters of α - hydroxy acid(s) were prepared via a direct nucleophilic substitution reaction. α - hydroxy acid(s) react with tosyl chloride in the presence of pyridine in acetonitrile at room temperature for overnight to give the desired compounds (compound I to III). The schematic route of

synthesis of *p*-toluenesulfonyl glycolate is given in **Scheme 1**. The base pyridine abstracts the acidic hydrogen from glycolic acid and produces the nucleophile (glycolate anion) which displaces chlorine atom in tosyl chloride to form the corresponding tosyl ester.

The structure of the synthesized tosyl ester(s) of  $\alpha$  - hydroxy acid(s) are characterized and confirmed by spectral measurements.

Antimicrobial activities on the synthesized compounds have been studied by Agar well diffusion method [23]. Antibacterial activities of the synthesized compounds were tested on gram positive *Staphylococcus aureus* and gram negative *Escherichia coli*, and antifungal activities were tested using *Aspergillus niger* and *Candida albicans*. The activities of the compounds were determined by measuring the zone of inhibition in mm around the test material and the results obtained were given in Table 1.

The results of the antimicrobial activity shows that the growth of *Staphylococcus aureus* is equally inhibited by both the compounds PTSG and PTSM. All the three esters synthesized show better activity against *Escherichia coli* and *Candida albicans*. PTSM shows excellent activity against *Aspergillus niger*.

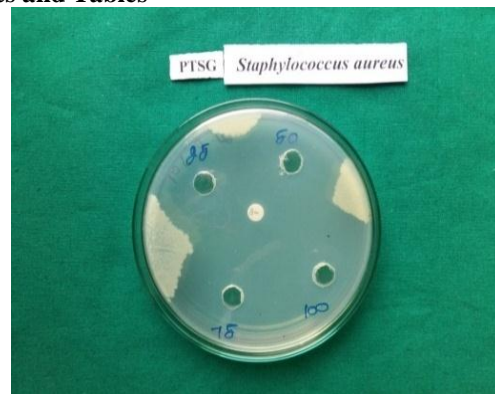
From the results, it is clear that the synthesized compounds produced much larger zones of inhibition against bacterial and fungal strains and infers that the synthesized compounds have more antimicrobial activities towards the microorganism with minimum inhibitory concentration when compared to standard antibiotics. The synthesized compounds are able to interfere with the metabolic process that are essential for the growth of the micro-organisms and hence inhibit their growth significantly. It means that the compounds are active against bacterial and fungal strains. The effect of concentration of the test samples has also been studied. The result shows that antimicrobial activity of the test samples increases with increase of their concentrations. As the concentration of the sample solution increases the diameter of the zone of inhibition also increases in case of all the compounds tested against both bacterial and fungal strains. The results highlight the medicinal importance and drug activities of these sulfur compounds.

## V. CONCLUSION AND FUTURE SCOPE

Tosyl ester(s) of  $\alpha$ - hydroxy acid(s) were synthesized and characterized by FT-IR,  $^1\text{H-NMR}$  and  $^{13}\text{C-NMR}$ . From the spectral data the structure of the compound(s) were established. The synthesized tosyl esters (I to III) were screened for their antimicrobial activities against two bacterial and two fungal strains using standard antibiotics. From the zone of inhibition value it is clear that the antimicrobial activity of the synthesized samples increases

with increase of their concentrations. The compounds (I to III) are highly active against both bacterial and fungal strains at minimum concentration. In future we have planned to conduct in-vivo studies on synthesized tosyl esters.

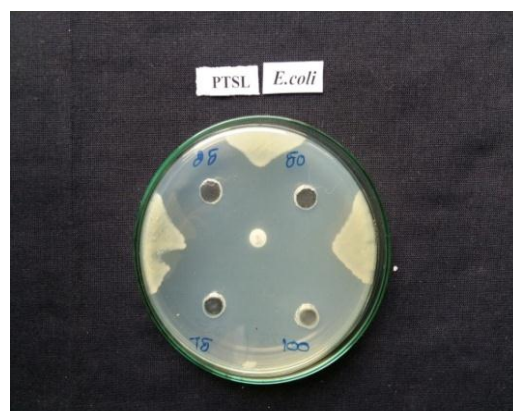
## Figures and Tables



**Figure 4: Antibacterial activity of *p*-toluenesulphonyl glycolate against *S. Aureus***



**Figure 5: Antibacterial activity of *p*-toluenesulphonyl glycolate against *A. niger***



**Figure 6: Antibacterial activity of *p*-toluenesulphonyl lactate against *E. coli***

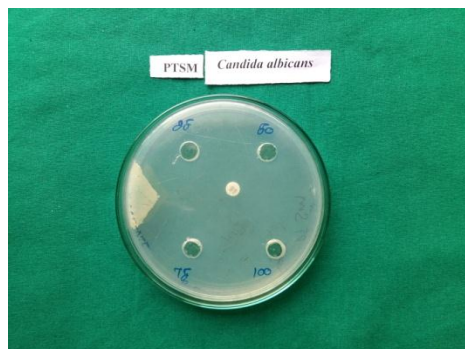


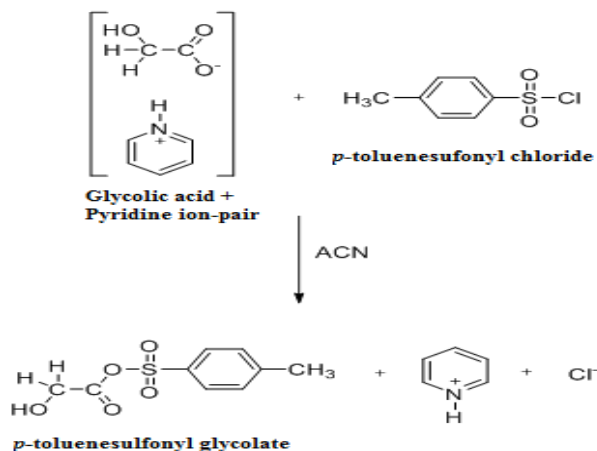
Figure 7: Antifungal activity of *p*-toluenesulphonyl mandalate against *C. albicans*

Table 1: Antimicrobial activity of tosyl ester(s) of  $\alpha$ -hydroxyacid(s)

Test sample	Micro organism tested	Zone of inhibition (mm) produced by different sample concentrations				
		25 $\mu\text{g/ml}$	50 $\mu\text{g/ml}$	75 $\mu\text{g/ml}$	100 $\mu\text{g/ml}$	Std.
I PTSG	<i>S. aureus</i>	40	42	46	50	28
	<i>E. coli</i>	36	38	41	44	15
	<i>C. albicans</i>	40	42	46	50	25
	<i>A. niger</i>	40	42	46	50	30
II PTSL	<i>S. aureus</i>	36	38	41	44	28
	<i>E. coli</i>	38	40	43	47	15
	<i>C. albicans</i>	40	42	46	50	25
	<i>A. niger</i>	40	42	46	50	30
III PTSM	<i>S. aureus</i>	40	43	46	50	28
	<i>E. coli</i>	38	40	43	47	15
	<i>C. albicans</i>	40	43	46	50	25
	<i>A. niger</i>	41	44	47	50	30

### STANDARD ANTIBIOTIC

Bacteria : Ciprofloxacin antibiotic disc  
*C. albicans* : Nystatin antibiotic disc  
*A. niger* : Amphotericin B antibiotic disc



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### REFERENCES

- [1] R. Ding, Y. He, X. Wang, J. Xu, Y. Chen, M. Feng and C. Qi, "Treatment of Alcohols with Tosyl Chloride Does Not always Lead to the Formation of Tosylates", *Molecules*, Vol. 16, pp. 5665 – 5673, 2011.
- [2] G.W. Kabalka, M. Varma, R.S. Varma, P.C. Srivastava, F.F. Knapp Jr., "The tosylation of alcohols", *J. Org. Chem.*, Vol. 51, Issue 12, pp. 2386-2388, 1986.
- [3] R. J. Cremlyn, 'An Introduction to organo Sulfur Chemistry', John Wiley & Sons; Chichester, pp-3, 219, 220, 222, 227, 233, 1996
- [4] G. L. Patrick, "An introduction to medicinal Chemistry", Oxford University Press, 2<sup>nd</sup> edition, pp -256, 2003.
- [5] W.H. Baarsche, "Reactions of organolithium compounds with sulfonate esters. A novel sulfone synthesis", *Can. J. Chem.*, Vol. 54, pp. 3056, 1976
- [6] R. S. Tipson, "On Esters of *p*-toluenesulfonic acid", *J. Org. Chem.*, Vol. 09, Issue 3, pp. 235 – 241, 1944.
- [7] C.H. Lee, S. D. Yoh, D. Y. Cheong, S. h. Kim and J.H. Park, "Product analysis in the reaction of substituted 1 – phenylethyl alcohols with *p*-toluenesulfonyl chloride", *Bull. Korean. Chem. Soc.*, Vol. 21, Issue 10, pp. 1049, 2000.
- [8] H. Gilman and N. J. Beaber, "The preparation of hydrocarbons by the reaction between alkyl sulfonates and organomagnesium halides", *J. Am. Chem. Soc.*, Vol. 47, Issue 2, pp.518 – 525, 1925.
- [9] S. Ozturk, H. Kutuk, "The Synthesis of Arylsulfonylphthalimides and Their Reactions with Several Amines in Acetonitrile", *International Journal of Organic Chemistry*, Vol.1, pp. 202-206, 2011
- [10] N. Vignola, S. Dahmen, D. Enders and S. Bra'se, "Synthesis of alkyl sulfonates from sulfonic acids or sodium sulfonates using solid-phase bound reagents", *Tetrahedron Letters*, Vol. 42 , pp. 7833–7836, 2001.
- [11] Y. A. Yusof & A. Ariffin, "Synthesis, Characterization and Antibacterial Activity of mono-, di- and tri-tosylate of Glycerol", *Sains Malaysiana*, Vol. 45, Issue 4, pp. 621–625, 2016.
- [12] J. W. Corcoran and F. E. Hahn, "Antibiotics, Volume III, Mechanism of Action of Antimicrobial and Antitumor Agents", Springer – Verlag Berlin. Heidelberg, New York, pp. 668 – 698, 1975.
- [13] M. E Wolff, "Burger's Medicinal Chemistry, Drug Discovery and Development, Therapeutic agents", V edn., John Wiley & Sons, New York, 1996.
- [14] J. Therese Punitha, S. Ananthalakshmi, M. Gowri and M. Anu, "Biological activity study on Indole acetic acid and its Cobalt(II), Nickel(II) and Copper(II) complexes", *Int. J. of Pharm. & Life Sci*, Vol.4, Issue 6, pp. 2746 – 2750, 2013.
- [15] S. Ananthalakshmi, R. Kavitha, A. Kiruthika, and S. Kalaivani, "Synthesis, Characterization and antimicrobial studies of tosyl esters of carboxylic acid", *International Journal of Scientific research Publications*", Vol.4, Issue 5, pp. 1 – 4, 2014.
- [16] A. I. Vogel, "A Text Book of Practical Organic Chemistry", 4<sup>th</sup> ed; ELBS; London, pp. 647, 1978.

- [17] D. D. Perrin & W. L. F. Armarego, "Purification of Laboratory Chemicals", 3<sup>rd</sup> edn., Pergamon Press, New York, pp. 189,210,214,267, 1988.
- [18] R. Kavitha and S. Ananthalakshmi, "Kinetic studies on the reaction of p-toluenesulfonyl chloride with  $\alpha$ -hydroxy acids in the presence of pyridine in acetonitrile", International Journal of Applied Research, Vol.2, Issue 11, pp. 317 – 321, 2016.
- [19] B. J. Wadher and G. L. Bhoosreddy, "Manual of Diagnostic microbiology", 1 Edn., Himalaya Publishing House, India, 1995.
- [20] M. Gowri, S. Ananthalakshmi and J. Therese Punitha, "In-Vitro antimicrobial screening of naphthalene acetic acid compounds", IJPLCP, Vol. 4, Issue 7, pp. 2780 -2784, 2013.
- [21] N. Sharma, "Antibacterial Activity of Fresh Juices of Lemon, Onion, Bottle Gourd and Tomato against Multiple Drug Resistant Bacteria such as E. Coli, Staphylococcus, Bacillus, Klebsiella and Salmonella", International Journal of Scientific Research in Biological Sciences, Vol.2, Issue 2, pp. 16-20, 2015.
- [22] R. Suresh, S. Thampiraj and A. Stephen "Antibacterial activities of wild rhizomatous plants – Curcuma aromatic, Curcuma longa (Zingiberaceae) and synergistic effects of both collected from southern western Ghats, India", International Journal of Scientific Research in Biological Sciences, Vol.5, Issue 2, pp. 7-13, 2018.
- [23] N. Pal, "Antibacterial activity of Hibiscus rosa sinensis and Calendula officinalis flowers extract against various pathogen", International Journal of Scientific Research in Biological Sciences, Vol.2, Issue 3, pp. 5-8, 2015.

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